

N-(4,6-Dimethylpyrimidin-2-yl)-1H-benzimidazol-2-amine

Shaaban Kamel Mohamed,^a Mahmoud A. A. El-Remaily,^b Atash V. Gurbanov,^c Ali N. Khalilov^c and Seik Weng Ng^{d*}

^aChemistry & Environmental Science Division, School of Science, Manchester Metropolitan University, England, ^bDepartment of Chemistry, Sohag University, Sohag, Egypt, ^cDepartment of Organic Chemistry, Baku State University, Baku, Azerbaijan, and ^dDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

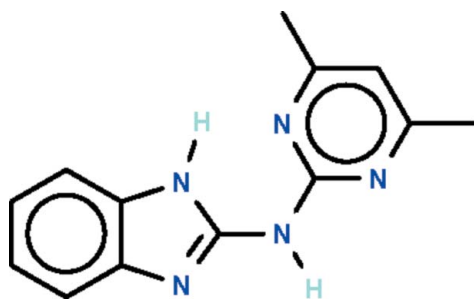
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Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.044; wR factor = 0.136; data-to-parameter ratio = 16.1.

There are two independent molecules in the asymmetric unit of the title compound, $\text{C}_{13}\text{H}_{13}\text{N}_5$. In each molecule, an amino N atom is connected to a benzimidazole fused-ring system and a pyrimidine ring [these are aligned at 1.3 (1)° in one independent molecule and at 5.4 (1)° in the other]. The amino N atom of the fused ring forms an intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond to a pyrimidine N atom in each molecule. The amino N atom connecting the two ring systems interacts with the other N atom of the pyrimidine ring of an adjacent molecule, generating centrosymmetric hydrogen-bonded dimers.

Related literature

For the synthesis, see: Bossio *et al.* (1985); Shestakov *et al.* (2006).



Experimental

Crystal data

$\text{C}_{13}\text{H}_{13}\text{N}_5$
 $M_r = 239.28$
 Triclinic, $P\bar{1}$
 $a = 8.2836$ (4) Å
 $b = 9.6135$ (5) Å
 $c = 16.5694$ (8) Å
 $\alpha = 92.121$ (1)°
 $\beta = 96.100$ (1)°
 $\gamma = 112.597$ (1)°
 $V = 1206.94$ (10) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 295$ K
 $0.30 \times 0.20 \times 0.20$ mm

Data collection

Bruker APEXII diffractometer
 13373 measured reflections
 5550 independent reflections
 4124 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$
 $wR(F^2) = 0.136$
 $S = 1.01$
 5550 reflections
 345 parameters
 4 restraints
 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.25$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{N5}$	0.86 (1)	2.05 (2)	2.664 (2)	128 (2)
$\text{N3}-\text{H3}\cdots\text{N2}^i$	0.87 (1)	2.05 (1)	2.912 (2)	170 (2)
$\text{N7}-\text{H7}\cdots\text{N10}$	0.87 (1)	2.10 (2)	2.695 (2)	125 (2)
$\text{N8}-\text{H8}\cdots\text{N6}^{ii}$	0.87 (1)	2.05 (1)	2.908 (2)	170 (2)

Symmetry codes: (i) $-x + 2, -y + 1, -z + 1$; (ii) $-x + 2, -y + 2, -z + 2$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5003).

References

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supplementary materials

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N-(4,6-Dimethylpyrimidin-2-yl)-1*H*-benzimidazol-2-amine

S. K. Mohamed, M. A. A. El-Remaily, A. V. Gurbanov, A. N. Khalilov and S. W. Ng

Comment

The reported synthesis involves the reaction of 4,6-dimethylpyrimidin-2-ylcyanamide with *o*-phenylenediamine, and it illustrates the type of heterocycles that are formed by the reaction of cyanamides with *N,N*-binucleophiles (Shestakov *et al.*, 2006). The unsubstituted compound was reported earlier (Bossio *et al.*, 1985). The present synthesis is a more convenient synthesis that uses acetylacetone as one of the reactants. An amino N atom in the approximately planar C₁₃H₁₃N₅ molecule is connected to a benzimidazolyl fused-ring and a pyrimidinyl ring; the amino N atom of the fused ring forms an intramolecular N–H···O hydrogen bond to a pyrimidinyl N atom (Scheme I, Fig. 1). There are two independent molecules; each molecule is connected to an inversion-related molecule by an N–H···O hydrogen bond.

Experimental

1*H*-Benzo[*d*]imidazol-2-yliminomethanediamine (0.05 mol) and acetylacetone (0.10 mol, approx. 10 ml) along with several drops of acetic acid were heated at 473 for 1 h. The solid that formed on cooling was collected and recrystallized from ethanol to give the title compound in 80% yield; m.p. 623 K.

Refinement

Carbon-bound H-atoms were placed in calculated positions [C–H 0.93 to 0.96 Å; *U*(H) 1.2 to 1.5*U*(C)] and were included in the refinement in the riding model approximation.

The amino H-atoms were located in a difference Fourier map, and were refined with a distance restraint of N–H 0.86±0.01 Å; their temperature factors were refined.

Figures

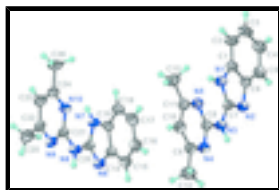


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of the two independent molecules of C₁₃H₁₃N₅ at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

N-(4,6-Dimethylpyrimidin-2-yl)-1*H*-benzimidazol-2-amine

Crystal data

C₁₃H₁₃N₅
M_r = 239.28

Z = 4
F(000) = 504

supplementary materials

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 8.2836$ (4) Å

$b = 9.6135$ (5) Å

$c = 16.5694$ (8) Å

$\alpha = 92.121$ (1)°

$\beta = 96.100$ (1)°

$\gamma = 112.597$ (1)°

$V = 1206.94$ (10) Å³

$D_x = 1.317$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 4428 reflections

$\theta = 2.3$ – 27.4 °

$\mu = 0.09$ mm⁻¹

$T = 295$ K

Prism, colorless

$0.30 \times 0.20 \times 0.20$ mm

Data collection

Bruker APEXII
diffractometer

Radiation source: fine-focus sealed tube

graphite

φ and ω scans

13373 measured reflections

5550 independent reflections

4124 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

$\theta_{\text{max}} = 27.5$ °, $\theta_{\text{min}} = 1.2$ °

$h = -10 \rightarrow 10$

$k = -12 \rightarrow 12$

$l = -21 \rightarrow 21$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.136$

$S = 1.01$

5550 reflections

345 parameters

4 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0736P)^2 + 0.1933P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.25$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.66435 (17)	0.61916 (15)	0.45010 (8)	0.0480 (3)
N2	0.80016 (16)	0.45944 (14)	0.43969 (8)	0.0470 (3)
N3	0.94151 (17)	0.67635 (15)	0.53228 (8)	0.0498 (3)
N4	1.11489 (17)	0.88346 (14)	0.61816 (8)	0.0491 (3)
N5	0.83527 (16)	0.86534 (14)	0.55242 (8)	0.0477 (3)
N6	0.98905 (15)	1.11808 (14)	0.92502 (8)	0.0453 (3)
N7	0.72463 (16)	1.09201 (14)	0.86230 (8)	0.0452 (3)
N8	0.74067 (16)	0.92463 (15)	0.96388 (8)	0.0485 (3)
N9	0.52809 (16)	0.73369 (14)	1.01630 (8)	0.0477 (3)
N10	0.44770 (16)	0.87329 (14)	0.91397 (8)	0.0464 (3)

C1	0.5574 (2)	0.50529 (17)	0.39234 (9)	0.0453 (3)
C2	0.3985 (2)	0.48064 (19)	0.34608 (11)	0.0570 (4)
H2	0.3415	0.5464	0.3516	0.068*
C3	0.3283 (2)	0.3542 (2)	0.29138 (11)	0.0648 (5)
H3A	0.2219	0.3343	0.2590	0.078*
C4	0.4132 (2)	0.2562 (2)	0.28378 (11)	0.0646 (5)
H4	0.3630	0.1726	0.2458	0.077*
C5	0.5706 (2)	0.27942 (19)	0.33104 (10)	0.0566 (4)
H5	0.6258	0.2122	0.3261	0.068*
C6	0.64326 (19)	0.40638 (17)	0.38613 (9)	0.0446 (3)
C7	0.80551 (19)	0.58625 (16)	0.47531 (9)	0.0433 (3)
C8	0.96246 (19)	0.81387 (17)	0.56914 (9)	0.0444 (3)
C9	1.1398 (2)	1.01836 (18)	0.65380 (10)	0.0502 (4)
C10	1.0141 (2)	1.07950 (19)	0.64206 (10)	0.0541 (4)
H10	1.0317	1.1723	0.6684	0.065*
C11	0.8619 (2)	0.99939 (18)	0.59025 (10)	0.0492 (4)
C12	1.3142 (2)	1.1021 (2)	0.70528 (13)	0.0679 (5)
H12A	1.3969	1.0598	0.6921	0.102*
H12B	1.2994	1.0930	0.7618	0.102*
H12C	1.3577	1.2069	0.6949	0.102*
C13	0.7188 (2)	1.0572 (2)	0.57189 (12)	0.0638 (5)
H13A	0.6895	1.0528	0.5140	0.096*
H13B	0.7591	1.1600	0.5946	0.096*
H13C	0.6163	0.9959	0.5953	0.096*
C14	1.01079 (19)	1.22562 (16)	0.86868 (9)	0.0423 (3)
C15	1.1636 (2)	1.33597 (18)	0.84877 (10)	0.0510 (4)
H15	1.2738	1.3468	0.8743	0.061*
C16	1.1476 (2)	1.42941 (19)	0.78990 (10)	0.0563 (4)
H16	1.2488	1.5041	0.7758	0.068*
C17	0.9839 (2)	1.41438 (19)	0.75126 (10)	0.0569 (4)
H17	0.9777	1.4789	0.7117	0.068*
C18	0.8299 (2)	1.30521 (19)	0.77048 (10)	0.0525 (4)
H18	0.7199	1.2954	0.7451	0.063*
C19	0.84684 (19)	1.21133 (16)	0.82922 (9)	0.0433 (3)
C20	0.81693 (18)	1.04241 (16)	0.91841 (9)	0.0417 (3)
C21	0.56341 (18)	0.84116 (16)	0.96416 (9)	0.0433 (3)
C22	0.3584 (2)	0.64850 (18)	1.01769 (10)	0.0489 (4)
C23	0.2285 (2)	0.67368 (19)	0.96860 (11)	0.0554 (4)
H23	0.1101	0.6147	0.9705	0.066*
C24	0.27647 (19)	0.78683 (18)	0.91699 (10)	0.0494 (4)
C25	0.3156 (2)	0.5228 (2)	1.07291 (12)	0.0647 (5)
H25A	0.4053	0.5504	1.1191	0.097*
H25B	0.3105	0.4326	1.0439	0.097*
H25C	0.2036	0.5048	1.0911	0.097*
C26	0.1440 (2)	0.8194 (2)	0.86077 (12)	0.0683 (5)
H26A	0.1696	0.9257	0.8660	0.102*
H26B	0.0280	0.7643	0.8746	0.102*
H26C	0.1495	0.7889	0.8056	0.102*
H1	0.653 (2)	0.6974 (15)	0.4721 (10)	0.062 (5)*

supplementary materials

H3	1.0279 (19)	0.647 (2)	0.5423 (11)	0.067 (5)*
H7	0.6096 (12)	1.0539 (19)	0.8546 (11)	0.063 (5)*
H8	0.811 (2)	0.902 (2)	0.9979 (9)	0.059 (5)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0468 (7)	0.0444 (7)	0.0547 (8)	0.0227 (6)	-0.0035 (6)	0.0018 (6)
N2	0.0440 (7)	0.0451 (7)	0.0521 (7)	0.0199 (6)	-0.0010 (6)	-0.0007 (6)
N3	0.0447 (7)	0.0479 (7)	0.0585 (8)	0.0243 (6)	-0.0064 (6)	-0.0042 (6)
N4	0.0465 (7)	0.0477 (7)	0.0525 (7)	0.0203 (6)	-0.0013 (6)	-0.0007 (6)
N5	0.0470 (7)	0.0495 (7)	0.0504 (7)	0.0239 (6)	0.0032 (6)	0.0015 (6)
N6	0.0374 (6)	0.0460 (7)	0.0527 (7)	0.0178 (5)	-0.0004 (5)	0.0092 (5)
N7	0.0379 (6)	0.0476 (7)	0.0497 (7)	0.0186 (6)	-0.0035 (5)	0.0055 (5)
N8	0.0357 (6)	0.0484 (7)	0.0606 (8)	0.0167 (5)	-0.0021 (6)	0.0146 (6)
N9	0.0400 (6)	0.0491 (7)	0.0567 (8)	0.0203 (6)	0.0054 (6)	0.0072 (6)
N10	0.0383 (6)	0.0493 (7)	0.0498 (7)	0.0179 (5)	-0.0032 (5)	-0.0003 (6)
C1	0.0457 (8)	0.0438 (8)	0.0443 (8)	0.0157 (6)	0.0015 (6)	0.0101 (6)
C2	0.0526 (9)	0.0554 (9)	0.0616 (10)	0.0228 (8)	-0.0072 (8)	0.0119 (8)
C3	0.0567 (10)	0.0670 (11)	0.0595 (10)	0.0173 (9)	-0.0133 (8)	0.0074 (9)
C4	0.0641 (11)	0.0588 (10)	0.0566 (10)	0.0128 (9)	-0.0057 (8)	-0.0046 (8)
C5	0.0571 (10)	0.0504 (9)	0.0582 (10)	0.0182 (8)	0.0030 (8)	-0.0021 (7)
C6	0.0438 (8)	0.0451 (8)	0.0438 (8)	0.0160 (6)	0.0047 (6)	0.0079 (6)
C7	0.0415 (7)	0.0440 (8)	0.0456 (8)	0.0189 (6)	0.0020 (6)	0.0058 (6)
C8	0.0442 (8)	0.0441 (8)	0.0458 (8)	0.0187 (6)	0.0041 (6)	0.0032 (6)
C9	0.0507 (9)	0.0489 (8)	0.0486 (8)	0.0178 (7)	0.0029 (7)	0.0017 (7)
C10	0.0586 (10)	0.0495 (9)	0.0560 (9)	0.0245 (8)	0.0055 (8)	-0.0047 (7)
C11	0.0523 (9)	0.0510 (8)	0.0501 (8)	0.0259 (7)	0.0095 (7)	0.0035 (7)
C12	0.0601 (11)	0.0584 (10)	0.0767 (12)	0.0205 (9)	-0.0115 (9)	-0.0104 (9)
C13	0.0617 (11)	0.0672 (11)	0.0729 (12)	0.0382 (9)	0.0051 (9)	-0.0016 (9)
C14	0.0429 (7)	0.0425 (7)	0.0427 (7)	0.0196 (6)	0.0004 (6)	0.0015 (6)
C15	0.0464 (8)	0.0503 (9)	0.0527 (9)	0.0161 (7)	0.0018 (7)	0.0046 (7)
C16	0.0619 (10)	0.0485 (9)	0.0538 (9)	0.0158 (8)	0.0094 (8)	0.0069 (7)
C17	0.0740 (11)	0.0530 (9)	0.0490 (9)	0.0308 (9)	0.0052 (8)	0.0115 (7)
C18	0.0580 (9)	0.0567 (9)	0.0482 (9)	0.0306 (8)	-0.0026 (7)	0.0052 (7)
C19	0.0472 (8)	0.0428 (8)	0.0418 (7)	0.0218 (6)	-0.0007 (6)	-0.0004 (6)
C20	0.0381 (7)	0.0419 (7)	0.0466 (8)	0.0193 (6)	-0.0016 (6)	0.0028 (6)
C21	0.0370 (7)	0.0434 (8)	0.0503 (8)	0.0181 (6)	0.0009 (6)	-0.0005 (6)
C22	0.0431 (8)	0.0494 (8)	0.0560 (9)	0.0194 (7)	0.0101 (7)	0.0004 (7)
C23	0.0370 (8)	0.0599 (10)	0.0646 (10)	0.0145 (7)	0.0051 (7)	0.0024 (8)
C24	0.0372 (7)	0.0552 (9)	0.0527 (9)	0.0176 (7)	-0.0023 (6)	-0.0058 (7)
C25	0.0547 (10)	0.0660 (11)	0.0781 (12)	0.0242 (9)	0.0201 (9)	0.0210 (9)
C26	0.0422 (9)	0.0854 (13)	0.0718 (12)	0.0228 (9)	-0.0082 (8)	0.0089 (10)

Geometric parameters (\AA , $^\circ$)

N1—C7	1.3535 (18)	C5—H5	0.9300
N1—C1	1.377 (2)	C9—C10	1.380 (2)
N1—H1	0.86 (1)	C9—C12	1.502 (2)

N2—C7	1.3176 (19)	C10—C11	1.376 (2)
N2—C6	1.3960 (19)	C10—H10	0.9300
N3—C7	1.3689 (19)	C11—C13	1.499 (2)
N3—C8	1.3756 (19)	C12—H12A	0.9600
N3—H3	0.87 (1)	C12—H12B	0.9600
N4—C9	1.336 (2)	C12—H12C	0.9600
N4—C8	1.3396 (19)	C13—H13A	0.9600
N5—C8	1.3343 (18)	C13—H13B	0.9600
N5—C11	1.340 (2)	C13—H13C	0.9600
N6—C20	1.3183 (18)	C14—C15	1.385 (2)
N6—C14	1.3917 (19)	C14—C19	1.397 (2)
N7—C20	1.3558 (18)	C15—C16	1.380 (2)
N7—C19	1.383 (2)	C15—H15	0.9300
N7—H7	0.87 (1)	C16—C17	1.388 (2)
N8—C20	1.3650 (19)	C16—H16	0.9300
N8—C21	1.3771 (19)	C17—C18	1.383 (2)
N8—H8	0.87 (1)	C17—H17	0.9300
N9—C22	1.3310 (19)	C18—C19	1.384 (2)
N9—C21	1.3367 (19)	C18—H18	0.9300
N10—C21	1.3352 (18)	C22—C23	1.380 (2)
N10—C24	1.3481 (19)	C22—C25	1.498 (2)
C1—C2	1.382 (2)	C23—C24	1.372 (2)
C1—C6	1.396 (2)	C23—H23	0.9300
C2—C3	1.378 (3)	C24—C26	1.496 (2)
C2—H2	0.9300	C25—H25A	0.9600
C3—C4	1.384 (3)	C25—H25B	0.9600
C3—H3A	0.9300	C25—H25C	0.9600
C4—C5	1.382 (2)	C26—H26A	0.9600
C4—H4	0.9300	C26—H26B	0.9600
C5—C6	1.386 (2)	C26—H26C	0.9600
C7—N1—C1	106.87 (13)	C9—C12—H12C	109.5
C7—N1—H1	120.2 (12)	H12A—C12—H12C	109.5
C1—N1—H1	132.8 (12)	H12B—C12—H12C	109.5
C7—N2—C6	104.12 (12)	C11—C13—H13A	109.5
C7—N3—C8	126.91 (13)	C11—C13—H13B	109.5
C7—N3—H3	115.8 (13)	H13A—C13—H13B	109.5
C8—N3—H3	117.0 (13)	C11—C13—H13C	109.5
C9—N4—C8	115.66 (13)	H13A—C13—H13C	109.5
C8—N5—C11	116.23 (13)	H13B—C13—H13C	109.5
C20—N6—C14	104.20 (12)	C15—C14—N6	129.94 (13)
C20—N7—C19	106.69 (12)	C15—C14—C19	119.88 (14)
C20—N7—H7	121.9 (12)	N6—C14—C19	110.18 (13)
C19—N7—H7	131.3 (12)	C16—C15—C14	118.09 (15)
C20—N8—C21	127.59 (13)	C16—C15—H15	121.0
C20—N8—H8	116.5 (13)	C14—C15—H15	121.0
C21—N8—H8	115.9 (12)	C15—C16—C17	121.51 (16)
C22—N9—C21	116.15 (13)	C15—C16—H16	119.2
C21—N10—C24	115.57 (14)	C17—C16—H16	119.2
N1—C1—C2	132.25 (15)	C18—C17—C16	121.25 (15)

supplementary materials

N1—C1—C6	105.39 (13)	C18—C17—H17	119.4
C2—C1—C6	122.37 (15)	C16—C17—H17	119.4
C3—C2—C1	116.96 (16)	C17—C18—C19	116.95 (15)
C3—C2—H2	121.5	C17—C18—H18	121.5
C1—C2—H2	121.5	C19—C18—H18	121.5
C2—C3—C4	121.27 (16)	N7—C19—C18	132.51 (14)
C2—C3—H3A	119.4	N7—C19—C14	105.17 (12)
C4—C3—H3A	119.4	C18—C19—C14	122.32 (15)
C5—C4—C3	121.77 (16)	N6—C20—N7	113.75 (13)
C5—C4—H4	119.1	N6—C20—N8	122.52 (13)
C3—C4—H4	119.1	N7—C20—N8	123.73 (13)
C4—C5—C6	117.67 (16)	N10—C21—N9	127.34 (13)
C4—C5—H5	121.2	N10—C21—N8	118.60 (14)
C6—C5—H5	121.2	N9—C21—N8	114.06 (12)
C5—C6—C1	119.94 (14)	N9—C22—C23	120.91 (15)
C5—C6—N2	130.14 (14)	N9—C22—C25	117.17 (14)
C1—C6—N2	109.91 (13)	C23—C22—C25	121.90 (15)
N2—C7—N1	113.70 (13)	C24—C23—C22	119.11 (14)
N2—C7—N3	123.06 (13)	C24—C23—H23	120.4
N1—C7—N3	123.24 (14)	C22—C23—H23	120.4
N5—C8—N4	126.91 (14)	N10—C24—C23	120.91 (14)
N5—C8—N3	118.65 (13)	N10—C24—C26	116.69 (15)
N4—C8—N3	114.42 (13)	C23—C24—C26	122.39 (15)
N4—C9—C10	121.78 (15)	C22—C25—H25A	109.5
N4—C9—C12	116.46 (14)	C22—C25—H25B	109.5
C10—C9—C12	121.74 (15)	H25A—C25—H25B	109.5
C11—C10—C9	118.16 (15)	C22—C25—H25C	109.5
C11—C10—H10	120.9	H25A—C25—H25C	109.5
C9—C10—H10	120.9	H25B—C25—H25C	109.5
N5—C11—C10	121.22 (14)	C24—C26—H26A	109.5
N5—C11—C13	116.37 (15)	C24—C26—H26B	109.5
C10—C11—C13	122.40 (15)	H26A—C26—H26B	109.5
C9—C12—H12A	109.5	C24—C26—H26C	109.5
C9—C12—H12B	109.5	H26A—C26—H26C	109.5
H12A—C12—H12B	109.5	H26B—C26—H26C	109.5
C7—N1—C1—C2	-179.49 (17)	C20—N6—C14—C15	-179.64 (15)
C7—N1—C1—C6	0.26 (16)	C20—N6—C14—C19	0.28 (16)
N1—C1—C2—C3	178.38 (17)	N6—C14—C15—C16	-179.89 (15)
C6—C1—C2—C3	-1.3 (2)	C19—C14—C15—C16	0.2 (2)
C1—C2—C3—C4	0.4 (3)	C14—C15—C16—C17	0.0 (2)
C2—C3—C4—C5	0.7 (3)	C15—C16—C17—C18	0.2 (3)
C3—C4—C5—C6	-0.9 (3)	C16—C17—C18—C19	-0.6 (2)
C4—C5—C6—C1	0.0 (2)	C20—N7—C19—C18	-179.50 (16)
C4—C5—C6—N2	-178.51 (16)	C20—N7—C19—C14	0.38 (16)
N1—C1—C6—C5	-178.65 (14)	C17—C18—C19—N7	-179.37 (15)
C2—C1—C6—C5	1.1 (2)	C17—C18—C19—C14	0.8 (2)
N1—C1—C6—N2	0.17 (17)	C15—C14—C19—N7	179.52 (13)
C2—C1—C6—N2	179.95 (14)	N6—C14—C19—N7	-0.42 (16)
C7—N2—C6—C5	178.13 (16)	C15—C14—C19—C18	-0.6 (2)

C7—N2—C6—C1	-0.53 (16)	N6—C14—C19—C18	179.48 (14)
C6—N2—C7—N1	0.72 (17)	C14—N6—C20—N7	-0.03 (17)
C6—N2—C7—N3	-179.09 (14)	C14—N6—C20—N8	179.56 (13)
C1—N1—C7—N2	-0.64 (18)	C19—N7—C20—N6	-0.23 (17)
C1—N1—C7—N3	179.17 (14)	C19—N7—C20—N8	-179.82 (14)
C8—N3—C7—N2	176.24 (14)	C21—N8—C20—N6	177.63 (14)
C8—N3—C7—N1	-3.6 (3)	C21—N8—C20—N7	-2.8 (3)
C11—N5—C8—N4	-1.6 (2)	C24—N10—C21—N9	0.1 (2)
C11—N5—C8—N3	179.66 (14)	C24—N10—C21—N8	179.65 (13)
C9—N4—C8—N5	0.2 (2)	C22—N9—C21—N10	0.6 (2)
C9—N4—C8—N3	179.02 (13)	C22—N9—C21—N8	-178.95 (13)
C7—N3—C8—N5	3.8 (2)	C20—N8—C21—N10	1.4 (2)
C7—N3—C8—N4	-175.12 (15)	C20—N8—C21—N9	-179.03 (14)
C8—N4—C9—C10	1.5 (2)	C21—N9—C22—C23	-1.1 (2)
C8—N4—C9—C12	-176.69 (15)	C21—N9—C22—C25	177.35 (14)
N4—C9—C10—C11	-1.8 (3)	N9—C22—C23—C24	1.0 (2)
C12—C9—C10—C11	176.30 (16)	C25—C22—C23—C24	-177.42 (16)
C8—N5—C11—C10	1.2 (2)	C21—N10—C24—C23	-0.3 (2)
C8—N5—C11—C13	-179.73 (14)	C21—N10—C24—C26	-179.39 (14)
C9—C10—C11—N5	0.4 (3)	C22—C23—C24—N10	-0.2 (2)
C9—C10—C11—C13	-178.64 (16)	C22—C23—C24—C26	178.81 (16)

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1 \cdots N5	0.86 (1)	2.05 (2)	2.664 (2)	128 (2)
N3—H3 \cdots N2 ⁱ	0.87 (1)	2.05 (1)	2.912 (2)	170 (2)
N7—H7 \cdots N10	0.87 (1)	2.10 (2)	2.695 (2)	125 (2)
N8—H8 \cdots N6 ⁱⁱ	0.87 (1)	2.05 (1)	2.908 (2)	170 (2)

Symmetry codes: (i) $-x+2, -y+1, -z+1$; (ii) $-x+2, -y+2, -z+2$.

Fig. 1

